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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=295 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.038$
$w R$ factor $=0.103$
Data-to-parameter ratio $=16.5$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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# 2,5-Bis(3,7-dichloroquinolin-8-yl)-1,3,4-oxadiazole 

In the title molecular structure, $\mathrm{C}_{20} \mathrm{H}_{8} \mathrm{Cl}_{4} \mathrm{~N}_{4} \mathrm{O}$, the $1,3,4-$ triazole ring is twisted with an r.m.s. deviation of $0.0035 \AA$. One of the quinolinyl substituents makes a dihedral angle of $55.8(1)^{\circ}$ with respect to the central ring, while the other is rotated by 71.7 (1) ${ }^{\circ}$; these twists are necessary to relieve steric crowding.

## Comment

The title compound, (I), is a diaryl-substituted 1,3,4-oxadiazole, an example of which is the previously determined crystal structure of 2,5-diphenyl-1,3,4-triazole (Kuznetsov et al., 1998; Franco et al., 2003). The synthesis of 1,3,4-oxadiazoles is readily achieved by the treatment of aromatic carboxylic acids with hydrazine dihydrochloride in a mixture of orthophosphoric acid, phosphorus pentoxide and phosphorus oxychloride (Bentiss \& Lagrenee, 1999). Unlike the phenyl derivatives, only a few quinolinyl analogs have been reported (Dabhi et al., 1992; Narayana et al., 2005; Zhang et al., 1989).

(I)

The title 1,3,4-triazole incorporates the quinclorac entity, which is used commercially as a potent herbicide (Grossmann, 1998), into its structure (Fig. 1). The 1,3,4-triazole ring is twisted with an r.m.s. deviation of $0.0035 \AA$. One of the quinolinyl substituents (containing atom N1) makes a dihedral angle of $55.8(1)^{\circ}$ with respect to the central ring, while the other quinolyl ring (containing atom N4) is rotated by 71.7 (1) ${ }^{\circ}$. These large twists arise from steric interactions; in contrast, the diphenyl analog is essentially planar (Kuznetsov et al., 1998; Franco et al., 2003).

## Experimental

To a mixture of quinclorac (3,7-dichloro-8-quinolinecarboxylic acid) ( $24.2 \mathrm{~g}, 0.1 \mathrm{~mol}$ ) and hydrazinium sulfate ( $6.5 \mathrm{~g}, 0.05 \mathrm{~mol}$ ), which was dissolved in a mixture of $85 \%$ orthophosphoric acid ( 27 ml ), were added phosphorus pentoxide ( $42.6 \mathrm{~g}, 0.3 \mathrm{~mol}$ ) and phosphorus

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oxychloride ( $46.0 \mathrm{~g}, 0.1 \mathrm{~mol}$ ). The viscous liquid was heated at 413 K for 4 h . The cooled mixture was poured on to crushed ice. Sodium hydroxide was added until the solution registered a neutral pH . The resulting white solid was collected, washed with water and then dried and purified by recrystallization from ethanol in about $80 \%$ yield (m.p. 508-509 K).

## Crystal data

## $\mathrm{C}_{20} \mathrm{H}_{8} \mathrm{Cl}_{4} \mathrm{~N}_{4} \mathrm{O}$

$M_{r}=462.10$
Orthorhombic, $P$ ecn
$a=14.1016$ (8) $\AA$
$b=9.8811$ (6) $\AA$
$c=27.016$ (2) $\AA$
$V=3764.4(4) \AA^{3}$

$$
\begin{aligned}
& Z=8 \\
& D_{x}=1.631 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \mu=0.65 \mathrm{~mm}^{-1} \\
& T=295(2) \mathrm{K} \\
& \text { Block, colorless } \\
& 0.27 \times 0.17 \times 0.16 \mathrm{~mm}
\end{aligned}
$$

## Data collection

Bruker APEX-II area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: none
4317 independent reflections
2953 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.037$
$\theta_{\text {max }}=27.5^{\circ}$
21631 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.038$
$w R\left(F^{2}\right)=0.103$
$S=1.02$
4317 reflections
262 parameters
H-atom parameters constrained

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0437 P)^{2}\right. \\
& \quad+1.1566 P] \\
& \text { where } P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.23 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.30 \mathrm{e}^{-3}
\end{aligned}
$$



Figure 1
ORTEPII (Johnson, 1976) plot of (I). Displacement ellipsoids are drawn at the $50 \%$ probability level and H atoms are drawn as spheres of arbitrary radii.

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