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Key indicators

Single-crystal X-ray study T = 295 KMean $\sigma(\text{C}-\text{C}) = 0.003 \text{ Å}$ R factor = 0.038 wR factor = 0.103 Data-to-parameter ratio = 16.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2,5-Bis(3,7-dichloroquinolin-8-yl)-1,3,4-oxadiazole

In the title molecular structure, $C_{20}H_8Cl_4N_4O$, the 1,3,4triazole ring is twisted with an r.m.s. deviation of 0.0035 Å. One of the quinolinyl substituents makes a dihedral angle of 55.8 (1)° with respect to the central ring, while the other is rotated by 71.7 (1)°; these twists are necessary to relieve steric crowding.

Comment

The title compound, (I), is a diaryl-substituted 1,3,4-oxadiazole, an example of which is the previously determined crystal structure of 2,5-diphenyl-1,3,4-triazole (Kuznetsov *et al.*, 1998; Franco *et al.*, 2003). The synthesis of 1,3,4-oxadiazoles is readily achieved by the treatment of aromatic carboxylic acids with hydrazine dihydrochloride in a mixture of orthophosphoric acid, phosphorus pentoxide and phosphorus oxychloride (Bentiss & Lagrenee, 1999). Unlike the phenyl derivatives, only a few quinolinyl analogs have been reported (Dabhi *et al.*, 1992; Narayana *et al.*, 2005; Zhang *et al.*, 1989).



The title 1,3,4-triazole incorporates the quinclorac entity, which is used commercially as a potent herbicide (Grossmann, 1998), into its structure (Fig. 1). The 1,3,4-triazole ring is twisted with an r.m.s. deviation of 0.0035 Å. One of the quinolinyl substituents (containing atom N1) makes a dihedral angle of 55.8 (1)° with respect to the central ring, while the other quinolyl ring (containing atom N4) is rotated by 71.7 (1)°. These large twists arise from steric interactions; in contrast, the diphenyl analog is essentially planar (Kuznetsov *et al.*, 1998; Franco *et al.*, 2003).

Experimental

To a mixture of quinclorac (3,7-dichloro-8-quinolinecarboxylic acid) (24.2 g, 0.1 mol) and hydrazinium sulfate (6.5 g, 0.05 mol), which was dissolved in a mixture of 85% orthophosphoric acid (27 ml), were added phosphorus pentoxide (42.6 g, 0.3mol) and phosphorus

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oxychloride (46.0 g, 0.1 mol). The viscous liquid was heated at 413 K for 4 h. The cooled mixture was poured on to crushed ice. Sodium hydroxide was added until the solution registered a neutral pH. The resulting white solid was collected, washed with water and then dried and purified by recrystallization from ethanol in about 80% yield (m.p. 508–509 K).

Z = 8

 $D_x = 1.631 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

 $\mu = 0.65 \text{ mm}^{-1}$

T = 295 (2) K

 $R_{\rm int} = 0.037$

 $\theta_{\rm max} = 27.5^{\circ}$

Block colorless

 $0.27 \times 0.17 \times 0.16 \; \mathrm{mm}$

4317 independent reflections

2953 reflections with $I > 2\sigma(I)$

 $w = 1/[\sigma^2(F_0^2) + (0.0437P)^2]$

where $P = (F_0^2 + 2F_c^2)/3$

+ 1.1566P]

 $(\Delta/\sigma)_{\rm max} = 0.001$

 $\Delta \rho_{\rm max} = 0.23 \text{ e } \text{\AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.30 \text{ e } \text{\AA}^{-3}$

Crystal data

 $\begin{array}{l} C_{20}H_8Cl_4N_4O\\ M_r = 462.10\\ Orthorhombic, Pbcn\\ a = 14.1016 \ (8) \ \mathring{A}\\ b = 9.8811 \ (6) \ \mathring{A}\\ c = 27.016 \ (2) \ \mathring{A}\\ V = 3764.4 \ (4) \ \mathring{A}^3 \end{array}$

Data collection

Bruker APEX-II area-detector diffractometer φ and ω scans Absorption correction: none 21631 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.038$ $wR(F^2) = 0.103$ S = 1.024317 reflections 262 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

O1-C11	1.354 (2)	N2-N3	1.420 (2)
O1-C10 N2 C10	1.358 (2)	N3-C11	1.282 (3)
$C_{11} = C_{10}$	1.280(3) 1030(2)	$N^2 - C^{10} - O^1$	112 4 (2)
C10-N2-N3	106.1 (2)	N3-C11-O1	112.7 (2)
C11-N3-N2	105.7 (2)		

H atoms were positioned geometrically (C-H = 0.93 and 0.97Å) and were included in the refinement in the riding-model approximation, with $U_{iso}(H)$ values set at $1.2U_{eq}(C)$.

Data collection: *SMART* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976) and *SHELXTL* (Bruker, 2004); software used to prepare material for publication: *SHELXL97*.



Figure 1

ORTEPII (Johnson, 1976) plot of (I). Displacement ellipsoids are drawn at the 50% probability level and H atoms are drawn as spheres of arbitrary radii.

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